

Reagent-Controlled Regiodivergent Resolution of Unsymmetrical Oxabicyclic Alkenes Using a Cationic Rhodium Catalyst

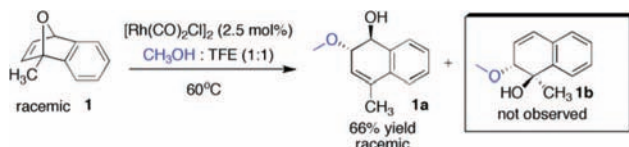
Robert Webster, Christian Böing, and Mark Lautens*

Department of Chemistry, University of Toronto, Toronto, Ontario, Canada, M5S 3H6

Received October 7, 2008; E-mail: mlautens@chem.utoronto.ca

Transition metal catalyzed reactions of oxa- and azabicyclic alkenes have proven extremely useful in the synthesis of materials ranging from biologically active molecules of pharmaceutical interest¹ to polymer components.² Many highly efficient asymmetric transformations involving these unique moieties have emerged in the literature³ as attractive methods to allow rapid access to difficult-to-make hydroxy-dihydronaphthalene cores. In our continuing effort to identify useful reactivity profiles in these systems, an enantiomerically pure bridgehead substituted oxabicyclic was synthesized⁴ and its ring opening chemistry with heteroatom nucleophiles was explored under rhodium catalysis. From our earlier work, it is known that a bridgehead substituted oxabenzonorbornene undergoes Rh-catalyzed addition of methanol with complete substrate control (Scheme 1).⁵

Scheme 1. Substrate Control in Rh-Catalyzed Ring Opening⁵

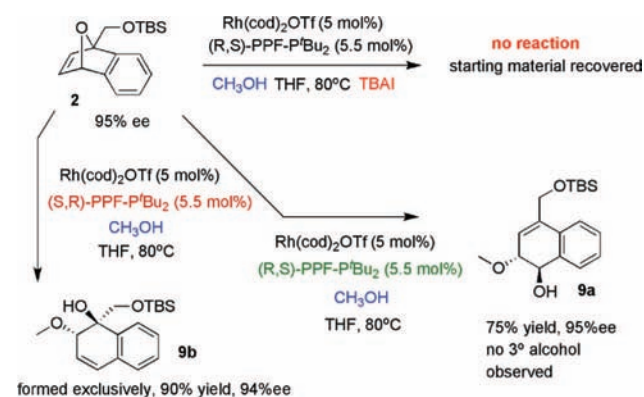


Josiphos type ligands complexed to Rh promote ring opening of meso oxabicyclic and azabicyclic alkenes with heteroatom nucleophiles with very high enantioselectivities. Conventionally, these transformations are carried out with Rh halide complexes, and earlier work in our group has shown that in situ generated Rh iodide complexes⁶ (via addition of tetrabutylammonium iodide to the reaction) commonly give the best yields and enantioselectivities in these types of reactions. However, when oxabicyclic **2** (Scheme 2) was subjected to our standard ring opening protocol, it was found to be essentially inert. We suspected a more electrophilic catalyst system would be required, and the reaction was attempted using a cationic Rh(I) triflate catalyst in the presence of the (*S,R*)-PPF-P^tBu₂ Josiphos ligand. Previously we found this complex to be too reactive. Inherent, efficient opening was observed, and moreover only **9b** was observed which is regioisomeric to our earlier studies with achiral ligands. Reacting **2** with the enantiomeric ligand yielded the regioisomer **9a** implying the ligand overrides the inherent preference of the substrate.

In light of the dominant role previously exerted by the substrate, racemic bridgehead substituted oxabicyclics were investigated (Scheme 3) and the enantiomers were found to give regioisomeric products implying strong catalyst control.⁷ The use of the cationic Rh catalyst has proven to be essential, as replacing the counterion by chloride or iodide does not allow access to **1b** (product type **B**), which arises from the unexpected mode of ring opening.

Our findings imply that a matched/mismatched effect during the oxidative insertion of the Rh ligand complex into the C–O bridgehead bond is operative and responsible for the observed stereoselectivity. (Scheme 4) In all cases, one of the regioisomers

Scheme 2. Reagent Controlled Ring Opening Using Cationic Rh(I)



is produced with very high ee, while the other is formed in lower selectivity. Because the reaction products have two stereocenters, one of which is fixed in the starting material, for one of the products to form in less than perfect ee requires that it originate from the opposite enantiomer of the starting substrate. In general, (Scheme 4) the examples studied give product type **B** as virtually a single enantiomer. This can occur when substrate “R” reacts with the chiral Rh–ligand complex in a matched fashion and breaks the C–O bond in agreement with substrate control. Product type **A** arises from nucleophilic addition via pathway **a**, and pathway **b** (leading to product type **B**) is inoperative. The “S” antipode of the starting material interacts with the chiral catalyst in a mismatched manner, and the inherent reactivity of the substrate participates to a minor extent. The substrate controlled pathway **a** leads to product **A** with a slight erosion of ee, while ligand control gives rise to product **B** in very high enantiopurity. This model conforms to the trend that the product formed with higher enantioselectivity generally gives a slightly lower yield. Monitoring reaction progress over time using chiral HPLC (entry 4, Table 1) has shown that product ee is constant throughout the reaction, and both enantiomers of the starting material are consumed at very similar rates.

It was found that the degree of enantioselectivity results primarily from the identity of the bridgehead substituent. The unsymmetrical

Scheme 3. Regiodivergent Resolution Resolution Using Cationic Rh(I)

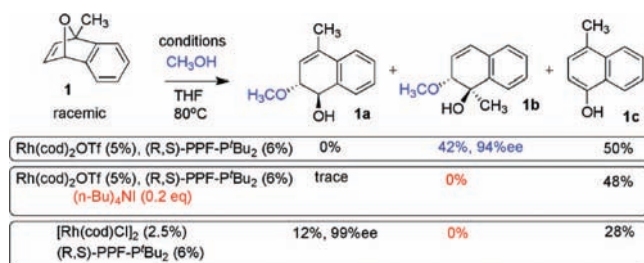
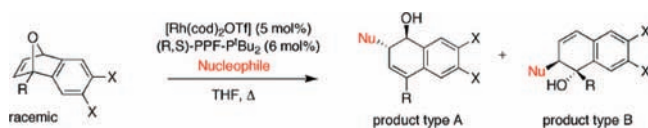
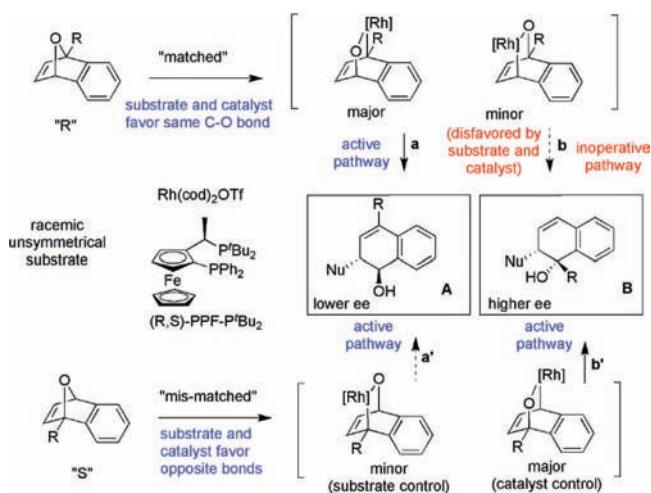
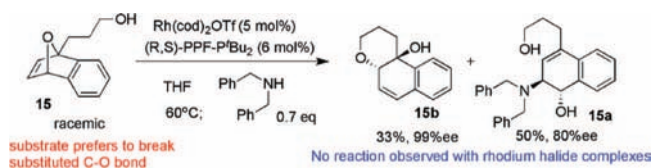


Table 1. Rh-Catalyzed Divergent Enantioselective Ring Opening

entry	R	X	nucleophile	product A		product B	
				yield (%)	ee ^d (%)	yield (%)	ee ^d (%)
1 ^b	CH ₃	H	CH ₃ OH	(50) ^a	-	42	90
2 ^b	CH ₃	H	Bn ₂ NH	29	86	32	>99
3 ^b	CH ₃	H	Et ₂ NH	48	79 ^c	35	>99
4 ^b	CH ₃	H	PhNHCH ₃	43	80	35	99
5 ^c	CH ₃	H		50	75	39	>99
6 ^c	CH ₃	Br	Bn ₂ NH	50	83	30	99
7 ^c	CH ₃	F	Bn ₂ NH	50	90	45	>99
8 ^{c,f}	CH ₃	F		50	90	48	>99
9 ^b	CH ₂ OTBS	H	CH ₃ OH	27	94	49	81
10 ^b	CH ₂ OTBS	H	Bn ₂ NH	27	99	32	74
11 ^b	CH ₂ Ph	H	Et ₂ NH	37	91 ^c	32	>99
12 ^b	CH ₂ Ph	H		47	82	36	>99
13 ^b	(CH ₂) ₂ CO ₂ Et	H	CH ₃ OH	41	>99	35	>99

^a Aromatized naphthol product **1c** was isolated. ^b Reaction performed at 80 °C. ^c Reaction performed at 60 °C. ^d Measured using HPLC analysis with chiral stationary phase columns; conditions are detailed in the Supporting Information. ^e HPLC analysis performed after conversion of the alcohol to the acetate. ^f Relative and absolute stereochemistry assigned by single crystal X-ray analysis.

compounds bearing a methyl group (entries 1–6, Table 1) give product type **A** in moderate to good ee and universally give product type **B** in very high ee. Changing the bridgehead group either leads to a reversal in which regioisomer is formed in high ee (entries 9 and 10, Table 1) or causes both type **A** and **B** products to be formed with very high stereoselectivity (entry 13, Table 1).

Scheme 4. Proposed Pathways Allowing for Observed Stereoselectivity**Scheme 5.** Regiodivergent Inter/Intramolecular Parallel Resolution

substrate prefers to break substituted C-O bond

No reaction observed with rhodium halide complexes

Tethering a nucleophile to the bridgehead position (Scheme 5) creates a situation wherein one enantiomer undergoes cyclization, whereas the other enantiomer is unable to do so resulting in decomposition. If an external nucleophile is added, then intermolecular opening of the complimentary enantiomer of substrate is observed.

In conclusion, we have demonstrated an unusual mode of reagent control using cationic Rh(I) complexes to resolve racemic bridgehead substituted oxabicyclic alkenes into pairs of regioisomeric products with high enantioselectivity.

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Supporting Information Available: Experimental procedures, characterization data, ¹H and ¹³C NMR spectra for new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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